- 1. CH<sub>4</sub>
- 2. H<sub>2</sub>O
- 3. NH<sub>3</sub>
- 4. HF
- 5. 2 single bonds and a double bond
- 6. degenerate orbitals (hybridized)
- 7. simplest saturated hydrocarbon
- 8. simplest unsaturated hydrocarbon
- 9. pi bonds (p overlaps) come from where? Give an example.
- 10.name the saturated hydrocarbon with a chain of 5 carbon atoms
- 11.conjugated systems come from where? Give an example.
- 12.BCl<sub>3</sub> does it hybridize?
- 13.H<sub>2</sub>S
- 14.CCl<sub>4</sub>
- 15.HBr
- 16,H<sub>2</sub>CO<sub>3</sub>
- 17.SF<sub>2</sub>
- 18.SiCl<sub>4</sub>
- 19.PH<sub>3</sub>
- 20.NCl<sub>3</sub>
- 21.C<sub>4</sub>H<sub>8</sub>
- $22.C_5H_{12}$
- 23.C<sub>4</sub>H<sub>6</sub>
- $24.C_{3}H_{6}$
- 25.CH<sub>3</sub>-CH=CH-CH<sub>2</sub>-CH<sub>3</sub>
- 26.SiF<sub>4</sub>
- $27.H_2Se$
- 28.CH<sub>3</sub>I
- 29.BeI<sub>2</sub>
- $30.BI_3$
- 31.CI<sub>4</sub>
- $32.OI_2$
- 33.hexane
- 34.cyclohexyne
- 35. cyclohexene