

# SCIENTIFIC MEASUREMENT

## **Practice Problems**

In your notebook, solve the following problems.

### SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY

Using different rulers, Bruce and Pete each measure the length of the same object three times.

- **1.** Bruce's three measurements are 19 cm, 20 cm, and 22 cm. Calculate the average value of his measurements and express the answer with the correct number of significant figures.
- **2.** Pete's three measurements are 20.9 cm, 21.0 cm, and 21.0 cm. Calculate the average value of his measurements and express the answer with the correct number of significant figures.
- **3.** Multiply the answer to problem 1 by the answer to problem 2. Express the answer in scientific notation with the correct number of significant figures.
- 4. Whose measurements are more precise?
- 5. The actual length of the object is 20 cm. Whose measurements are more accurate?
- 6. What is the error of Pete's average measurement?
- 7. What is the percent error of Pete's average measurement?
- **8.** Four boards each measuring 1.5 m are laid end to end. Multiply to determine the combined length of the boards, expressed with the correct number of significant figures.

#### **SECTION 3.2 THE INTERNATIONAL SYSTEM OF UNITS (SI)**

A fish tank measures 0.40 meter long by 0.20 meter wide by 0.30 meter high.

- 1. What is the width of the tank in centimeters?
- 2. What is the length of the tank in millimeters?
- **3.** What is the volume of the tank in liters?
- 4. What is the mass of water, in grams, that would fill the tank halfway? (1 L  $H_2O = 1$  kg)
- 5 How many nanoseconds are there in one minute?
- 6. A chemical reaction takes place at 20°C. What is this temperature in kelvins?
- 7. A typical refrigerator keeps food at 277 K. What is this temperature in degrees Celsius?

Material	Density at 20 <sup>o</sup> C (g/L)
Gold	19.3
Air	1.20
Water $(4^{0}C)$	1.000
Ice $(0^{0}C)$	0.917
Carbon dioxide	1.83
Aluminum	2.70

#### SECTION 3.4 DENSITY – Use the data above to solve problems 1-4.

1. What is the mass at 20°C of 5 liters of air?

**2.** A balloon filled with air is released in a room filled with carbon dioxide. Will the balloon float to the ceiling or sink to the floor?

**3.** What is the volume in liters of a kilogram of ice at  $0^{\circ}$ C?

4. What is the mass of a bar of aluminum measuring 1.0 cm by 1.0 cm by 10.0 cm?