Unit 9 – Chapter 15: Strong Acid-Strong Base Titration	Name
	Period
A 25.0-mL sample of 0.100 M HNO3 is titrated with 0.100 M NaOH. Calculate the p the following volumes of NaOH. Plot the results of your calculations as pH versus m the back of this sheet. 1) 0.0 mL of NaOH 7) 2) 4.0 mL of NaOH 8) 3) 8.0 mL of NaOH 9) 4) 12.5 mL of NaOH 10 5) 20.0 mL of NaOH 11	-
1) 0.0 mL of NaOH	7) 24.5 mL of NaOH
2) 4.0 mL of NaOH	8) 24.9 mL of NaOH
3) 8.0 mL of NaOH	9) 25.0 mL of NaOH
	10) 25 1 ml of No Oli
4) 12.5 mL of NaOH	10) 25.1 mL of NaOH
5) 20.0 mL of NaOH	11) 26.0 mL of NaOH
6) 24.0 mL of NaOH	12) 28.0 mL of NaOH

13) 30.0 mL of NaOH

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