

56. Three students made multiple weighings of a copper cylinder, each using a different balance. Describe the accuracy and precision of each student's measurements if the correct mass of the cylinder is 47.32 grams.

|            | Mass of cylinder (g) |        |        |
|------------|----------------------|--------|--------|
|            | Colin                | Lamont | Kivrin |
| Weighing 1 | 47.13                | 47.45  | 47.95  |
| Weighing 2 | 47.94                | 47.39  | 47.91  |
| Weighing 3 | 46.83                | 47.42  | 47.89  |
| Weighing 4 | 47.47                | 47.41  | 47.93  |

57. How many significant figures are in each underlined measurement?

- 60 seconds = 1 minute
- 47.70 grams of copper
- 1 km = 1000 meters
- 25 computers
- 9 innings in a baseball game
- 0.0950 meters of gold chain

58. Round off each of these measurements to three significant figures.

- 98.473 L
- 0.00076321 cg
- 57.048 m
- 12.17<sup>0</sup>C
- 0.0074983 X 10<sup>4</sup> mm
- 1764.9 mL

59. Round off each of the answers correctly

- 8.7 g + 15.43 + 19 g = 43.13 g
- 4.32 cm X 1.7 cm = 7.344 cm<sup>2</sup>
- 853.2 L – 627.443 L = 225.757 L
- 38.742 m<sup>2</sup> / 0.421 m = 92.023 m
- 5.40 m X 3.21 m X 1.871 m = 32.431914 m<sup>3</sup>
- 5.47 m<sup>3</sup> + 11 m<sup>3</sup> + 87.300 m<sup>3</sup> = 103.770 m<sup>3</sup>

60. Express each of the rounded-off answers in problems 58 & 59 in scientific notation

62. Write the SI base unit of measurement for each of these quantities

a. time

b. length

c. temperature

d. mass

e. energy

f. amount of a substance

63. Order these units from smallest to largest: cm,  $\mu\text{m}$ , km, mm, m, nm, dm, pm. Then give each measurement in terms of meters.

65. State the relationship between degrees Celsius and kelvins.

66. The melting point of silver is  $962^{\circ}\text{C}$ . Express this temperature in kelvins.

68. Would the density of a person be the same on the surface of Earth and on the surface of the moon? Explain.

69. A shiny, cold-colored bar of metal weighing 57.3 grams has a volume of  $4.7\text{ cm}^3$ . Is the bar of metal pure gold? (The density of gold is  $19.3\text{ g/cm}^3$ ).

70. Three balloons filled with neon, carbon dioxide, and hydrogen are released into the atmosphere. Describe the movement of each balloon if their densities are  $0.84\text{ g/cm}^3$ ,  $1.83\text{ g/cm}^3$ , and  $0.084\text{ g/cm}^3$  respectively. The density of air is  $1.20\text{ g/cm}^3$ .