## Chemistry – Chapter 3 Book assignment #3: Accuracy & precision, significant figures, metric system, density

1. Three students made multiple weighings of a copper cylinder, each using a different balance. Describe the accuracy and precision of each student's measurements if the correct mass of the cylinder is 47.32 grams.

	Mass of cylinder (g)		
	Colin	Lamont	Kivrin
Weighing 1	47.13	47.45	47.95
Weighing 2	47.94	47.39	47.91
Weighing 3	46.83	47.42	47.89
Weighing 4	47.47	47.41	47.93

- 2. How many significant figures are in each underlined measurement?
  - a. 60 seconds = 1 minute
  - b. 47.70 grams of copper
  - c. 1 km = <u>1000 meters</u>
  - d. 25 computers
  - e. 9 innings in a baseball game
  - f. 0.0950 meters of gold chain
- 3. Round off each of these measurements to three significant figures.
  - a. 98.473 L
  - b. 0.00076321 cg
  - c. 57.048 m
  - d. 12.17°C
  - e. 0.0074983 X 10<sup>4</sup> mm
  - f. 1764.9 mL
- 4. Round off each of the answers correctly
  - a. 8.7 g + 15.43 + 19 g = 43.13 g
  - b.  $4.32 \text{ cm } X 1.7 \text{ cm} = 7.344 \text{ cm}^2$
  - c. 853.2 L 627.443 L = 225.757 L
  - d.  $38.742 \text{ m}^2 / 0.421 \text{ m} = 92.023 \text{ m}$
  - e. 5.40 m X 3.21 m X 1.871 m = 32.431914 m<sup>3</sup>

- f.  $5.47 \text{ m}^3 + 11 \text{ m}^3 + 87.300 \text{ m}^3 = 103.770 \text{ m}^3$
- 5. Express each of the rounded-off answers in problems 3 & 4 in scientific notation
- 6. Write the SI base unit of measurement for each of these quantities
  - a. time
  - b. length
  - c. temperature
  - d. mass
  - e. energy
  - f. amount of a substance
- 7. Order these units from smallest to largest: cm, um, km, mm, m, nm, dm, pm. Then give each measurement in terms of meters.
- 8. State the relationship between degrees Celsius and kelvins.
- 9. The melting point of silver is 962°C. Express this temperature in kelvins.
- 10. Would the density of a person be the same on the surface of Earth and on the surface of the moon? Explain.
- 11. A shiny, cold-colored bar of metal weighing 57.3 grams has a volume of 4.7 cm<sup>3</sup>. Is the bar of metal pure gold? (The density of gold is 19.3 g/cm<sup>3</sup>).
- 12. Three balloons filled with neon, carbon dioxide, and hydrogen are released into the atmosphere Describe the movement of each balloon if their densities are 0.84 g/cm³, 1.83 g/cm³, and 0.084 g/cm³ respectively. The density of air is 1.20 g/cm³.